

Appl. No. 09/851,177

Amdt. dated September 22, 2003, and further amended February 6, 2007 and May 8, 2007

Reply to Office Action of March 26, 2003, and Office Actions of Jan. 9, 2007, Jan. 11, 2007, and Apr. 30, 2007

Amendments to the Claims:

This listing of claims will replace all prior versions, and listings, of claims in the application:

Listing of Claims:

Claim 1. (currently amended) A catalyst for use in the Fischer-Tropsch process, said catalyst comprising a catalyst particle, which comprises at least one metal that is an efficient carbon monoxide adsorber and at least one promoter, ~~said metal and said promoter being~~ dispersed on a support to form a said catalyst particle, said particle having a BET surface area of from about 100 m²/g to about 250 m²/g, and ~~said metal and said promoter being~~ are dispersed on the support such that the crystallite size of the metal oxide is from about 40 Å to about 200 Å, and said particle having an essentially smooth, homogeneous surface morphology.

Claim 2. (original) The catalyst of Claim 1 wherein said particle comprises from about 5 wt % to about 60 wt % cobalt, and from about 0.0001 wt % to about 1 wt % of a first promoter, and from about 0.01 wt % to about 5 wt % of a second promoter.

Claim 3. (original) The catalyst of Claim 2 wherein said particle comprises from about 10 wt% to about 30 wt % cobalt, and from about 0.01 wt % to about 0.05 wt % of said first promoter, and from about 0.1 wt % to about 1 wt % of said second promoter.

Claim 4. (original) The catalyst of Claim 1 wherein said metal is selected from the group consisting of nickel, cobalt, iron, ruthenium, osmium, platinum, palladium, iridium, rhenium, molybdenum, chromium, tungsten, vanadium, rhodium, copper, zinc, and combinations thereof.

Claim 5. (original) The catalyst of Claim 4 wherein said metal is cobalt.

Claim 6. (currently amended) The catalyst of Claim 1 wherein said promoter is selected from the group consisting of boron, cerium, chromium, copper, iridium, iron, lanthanum, manganese, molybdenum, palladium, platinum, rhenium, rhodium, ruthenium, strontium, tungsten, vanadium, zinc, sodium oxide, potassium oxide, rubidium oxide, cesium oxide, magnesium oxide, titanium oxide, zirconium oxide, ~~and other rare earth metals, such as~~ scandium, yttrium, praseodymium, neodymium, promethium, samarium, europium, gadolinium, terbium, dysprosium, holmium, erbium, thulium, ytterbium, lutetium, other rare earth metals and combinations thereof.

Claim 7. (original) The catalyst of Claim 2 wherein said first promoter is selected from the group consisting of palladium, platinum, ruthenium, rhenium, rhodium, iridium and a combination thereof; and said second promoter is selected from the group consisting of potassium, boron, cesium, lanthanum, cerium, strontium, scandium, yttrium, praseodymium, neodymium, promethium, samarium, europium, gadolinium, terbium, dysprosium, holmium, erbium, thulium, ytterbium, lutetium, palladium, platinum, ruthenium, rhenium, rhodium, iridium and combinations thereof.

Claim 8. (original) The catalyst of Claim 1 wherein said support is selected from the group consisting of aluminum oxide, γ-alumina, alumina monohydrate, alumina trihydrate, alumina-silica, magnesium silicate, silica, silicate, silicalite, y-zeolite, mordenite, titania, thoria, zirconia, niobia, hydrotalcite, kieselguhr, attapulgite clay, zinc oxide, other clays, other zeolites and combinations thereof.

Claim 9. (original) The catalyst of Claim 8 wherein said support is γ-alumina.

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Claim 10. (original) The catalyst of Claim 9 wherein said support has a particle size of from about 60 μm to about 150 μm , a surface area of from about 90 m^2/g to about 210 m^2/g , a pore volume of from about 0.35 ml/g to about 0.50 ml/g , and a pore diameter of from about 8 nm to about 20 nm.

Claim 11. (currently amended) A catalyst for use in the Fischer-Tropsch process, said catalyst comprising cobalt dispersed on a support to form a catalyst particle having a high surface area, a smooth, homogeneous surface morphology, an essentially uniform distribution of cobalt throughout the support and a small metal oxide crystallite size, and wherein said particle being is formed by the steps of:

- a) adding said support to water, with agitation, to form a slurry, and maintaining a slurry temperature at from about 35°C to about 210°C;
- b) adding an aqueous cobalt salt solution having a pH value greater than the point of zero charge of said support to said slurry with agitation and while maintaining said slurry temperature at from about 65°C to about 120°C;
- c) agitating said slurry and maintaining said slurry temperature at from about 65°C to about 120°C until said cobalt salt is ~~essentially completely~~ reacted with said support;
- d) separating said slurry into a solid portion and a liquor portion;
- e) washing said solid portion with water;
- f) drying and calcining said solid portion at from about 90°C to about 375°C to form catalyst particles; and
- g) reducing said catalyst particles by heating said particles from ambient temperature to from about 300°C to about 500°C at a rate of from about 0.1°C/min to about 10°C/min over a period of from about 5 hours to about 40 hours.

Claim 12. (original) The catalyst of Claim 11 wherein said support is selected from the group consisting of aluminum oxide, γ -alumina, alumina monohydrate, alumina trihydrate, alumina-silica, magnesium silicate, silica, silicate, silicalite, γ -zeolite, mordenite, titania, thoria, zirconia, niobia, hydrotalcite, kieselguhr, attapulgite clay, zinc oxide, other clays, other zeolites and combinations thereof.

Claim 13. (original) The catalyst of Claim 12 wherein said support is aluminum oxide.

Claim 14. (original) The catalyst of Claim 12 wherein said support has a particle size of from about 60 μm to about 150 μm , a surface area of from about 90 m^2/g to about 210 m^2/g , a pore volume of from about 0.35 ml/g to about 0.50 ml/g , and a pore diameter of from about 8 nm to about 20 nm.

Claim 15. (original) The catalyst of Claim 11 wherein said cobalt salt solution comprises water and a cobalt (II) complex having coordination sphere ligands selected from the group consisting of water, chloride ion, ammonia, pyridine, triphenylphosphine, 1,2-diaminoethane, diethylenetriamine, triethylenetetraamine, acetate, oxalate, 2,4-pentanedione, ethylenedinitilo tetraacetic acid, and combinations thereof.

Claim 16. (currently amended) The catalyst of Claim 15 wherein said cobalt (II) complex has coordination sphere ligands selected from the group consisting of water ~~molecules~~, ammonia, pyridine, diaminoethane, diethylenetriamine, triethylenetetraamine, and a combination thereof.

Claim 17. (original) The catalyst of Claim 16 wherein said cobalt (II) complex is hexaammine cobalt (II) carbonate.

Claim 18. (original) The catalyst of Claim 11 wherein the slurry temperature is maintained at from about 65°C to about 120°C in step a).

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Claim 19. (original) The catalyst of Claim 11 wherein said solid portion is dried at from about 120°C to about 260°C in step f).

Claim 20. (original) The catalyst of Claim 11 wherein said particles are reduced in step g) by heating said particles from ambient temperature to about 350°C at a rate of about 1.0°C/min and then holding said particles at about 350°C for from about 12 hours to about 16 hours.

Claim 21. (original) The catalyst of Claim 11 wherein said particles are further stabilized to prevent pyrophoric reactions when said particles are in the presence of air.

Claim 22. (currently amended) The catalyst of Claim 21 wherein said particles are stabilized by ~~being coated~~ coating with oil.

Claim 23. (currently amended) The catalyst of Claim 11 further including at least one promoter; wherein said promoter ~~being~~ is added with said cobalt salt solution.

Claim 24. (original) The catalyst of Claim 23 wherein said promoter is a metal salt selected from the group consisting of rhenium (VII) oxide, ruthenium nitrosyl nitrate, platinum chloride, platinum ammine nitrate, platinum ammine chloride, and combinations thereof.

Claim 25. (original) The catalyst of Claim 11 further including at least one promoter impregnated onto said catalyst particle after said particle is dried in step f), said promoter being impregnated onto said particle by dipping said particle in an aqueous solution of said promoter while maintaining agitation, and then separating said impregnated particles from said solution, and drying said impregnated particles.

Claims 26 – 34. (canceled)

Claim 35. (new) A catalyst for use in the Fischer-Tropsch process, said catalyst comprising cobalt dispersed on a support to form a catalyst particle, wherein said particle is formed by the steps of:

- a) adding said support to water, with agitation, to form a slurry, and maintaining a slurry temperature at from about 35°C to about 210°C;
- b) adding an aqueous cobalt salt solution having a pII value greater than the point of zero charge of said support to said slurry with agitation and while maintaining said slurry temperature at from about 65°C to about 120°C;
- c) agitating said slurry and maintaining said slurry temperature at from about 65°C to about 120°C until said cobalt salt is reacted with said support;
- d) separating said slurry into a solid portion and a liquor portion;
- e) washing said solid portion with water;
- f) drying and calcining said solid portion at from about 90°C to about 375°C to form catalyst particles;
- g) reducing said catalyst particles by heating said particles from ambient temperature to from about 300°C to about 500°C at a rate of from about 0.1°C/min to about 10°C/min over a period of from about 5 hours to about 40 hours; and
- h) stabilizing said particles to prevent pyrophoric reactions when said particles are in the presence of air by coating said particles with oil.

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Claim 36. (new) The catalyst of Claim 35 wherein said support is selected from the group consisting of aluminum oxide, γ -alumina, alumina monohydrate, alumina trihydrate, alumina-silica, magnesium silicate, silica, silicate, silicalite, γ -zeolite, mordenite, titania, thoria, zirconia, niobia, hydrotalcite, kieselguhr, attapulgite clay, zinc oxide, other clays, other zeolites and combinations thereof.

Claim 37. (new) The catalyst of Claim 36 wherein said support has a particle size of from about 60 μm to about 150 μm , a surface area of from about 90 m^2/g to about 210 m^2/g , a pore volume of from about 0.35 ml/g to about 0.50 ml/g , and a pore diameter of from about 8 nm to about 20 nm.

Claim 38. (new) The catalyst of Claim 35 wherein said cobalt salt solution comprises water and a cobalt (II) complex having coordination sphere ligands selected from the group consisting of water, chloride ion, ammonia, pyridine, triphenylphosphine, 1,2-diaminoethane, diethylenetriamine, triethylenetetraamine, acetate, oxalate, 2,4-pentanedione, ethylenedinitilo tetraacetic acid, and combinations thereof.

Claim 39. (new) The catalyst of Claim 38 wherein said cobalt (II) complex has coordination sphere ligands selected from the group consisting of water, ammonia, pyridine, diaminoethane, diethylenetriamine, triethylenetetraamine, and a combination thereof.

Claim 40. (new) The catalyst of Claim 35 wherein the slurry temperature is maintained at from about 65°C to about 120°C in step a).

Claim 41. (new) The catalyst of Claim 35 wherein said solid portion is dried at from about 120°C to about 260°C in step f).

Claim 42. (new) The catalyst of Claim 35 wherein said particles are reduced in step g) by heating said particles from ambient temperature to about 350°C at a rate of about 1.0°C/min and then holding said particles at about 350°C for from about 12 hours to about 16 hours.

Claim 43. (new) The catalyst of Claim 35 further including at least one promoter wherein said promoter is added with said cobalt salt solution.

Claim 44. (new) The catalyst of Claim 43 wherein said promoter is a metal salt selected from the group consisting of cobalt, iron, nickel, copper, zinc, manganese, and combinations thereof.

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